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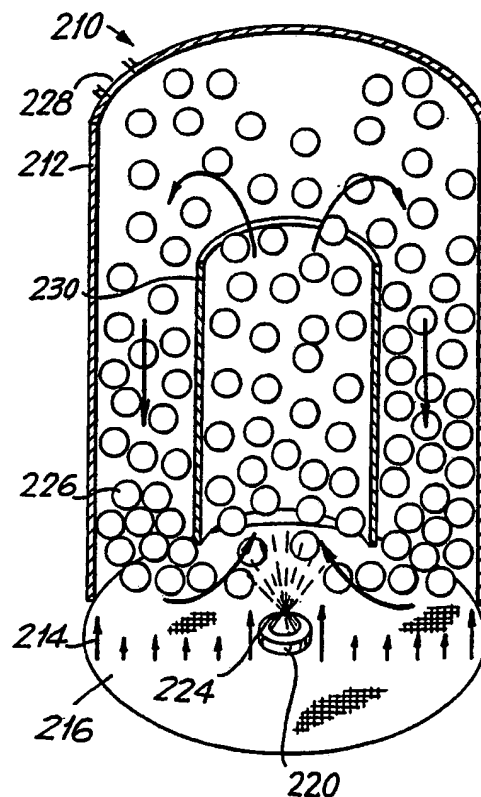
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(54) Title: STABLE AQUEOUS DRUG SUSPENSIONS**(57) Abstract**

Finely divided drug particles are coated with a lipid or bioadhesive polymer to form microspheres (226) having a particle size of about 550 micrometers or less, and coated with two or more enteric coatings, at least one of which is water insoluble, to form microcapsules. The resultant microcapsules can be suspended in an aqueous solution to form stable oral doses of the drug.



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STABLE AQUEOUS DRUG SUSPENSIONS

This invention relates to aqueous suspensions of encapsulated drugs having improved stability and process for making such suspensions.

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BACKGROUND OF THE INVENTION

Drugs such as amoxicillin, ampicillin, penicillin V and erythromycin are antibacterial drugs which are available for oral dispensation in a gelatin capsule containing a specified dosage amount of the drug. For patients who have difficulty swallowing capsules, e.g., the very young and the very old, the drugs can be suspended in an aqueous solution, such as a sugar-type syrup. However, the drugs are quite unstable in water, even when stored at temperatures of about 4°C, and very unstable at room temperature. Thus the drug solutions or suspensions have a very short shelf life, even at low temperatures. Further, they have an unpleasant taste which makes them unpalatable.

Beta-lactam antibiotics are orally inactive, and must be combined with an enhancer to promote their absorption into the body of the patient. Such enhancers are known, for example see US Patent 4,525,339, and include aliphatic fatty acids or acid glycerides. The fatty acids are generally C₂ to C₁₈ fatty acids, which can be straight or branched chain, saturated or unsaturated, their mono-, di- or triglycerides or mixtures thereof, and can also be partial or total esters of propylene

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glycol, polyethylene glycol and carbohydrates of C₂ to C₁₂ fatty acids and pharmaceutically acceptable esters and ethers of said glycerides. Encapsulating the drug in the enhancer is known also.

5 It would be desirable to be able to formulate these drugs so that they would be stable to long term storage, i.e., for up to about 18 months, even at room temperature in liquid form. In order to do that, the drug would have to be coated uniformly and completely with a water-impervious coating. Further, in
10 order for the drug to stay suspended in an aqueous solution or emulsion for administration in liquid form, the particle size of the coated drug particles would have to be very small, on the order of 1500 micrometers or less; otherwise the drug particles would settle out of the solution or suspension, and
15 the dosage would be inaccurate.

 A suitable coating or encapsulant material must be impervious to water, but must dissolve in the stomach or other appropriate portion of the digestive tract, depending on the drug to be administered and the dosage required, for absorption
20 by the patient. Coatings having particular solubility characteristics can be used to provide controlled or delayed release of the drug in the patient.

 Unfortunately, no one encapsulant coating is known to date that is effective to carry out all of these objectives. Thus
25 there is a need to be able to apply more than one encapsulant

material, successively and uniformly, over very small particle size granules of the drug to be administered. The preparation of small microspheres, i.e. having a particle size less than about 500 micrometers, microspheres containing the drug and
5 then encapsulating them in a series of encapsulants, that would provide enhancement, taste masking, controlled release and protection from moisture would be highly desirable.

SUMMARY OF THE INVENTION

We have found that microspheres comprising drug particles
10 which have a particle size of up to about 550 microns, coated with a matrix material of a lipid or a bioadhesive polymer, can be encapsulated uniformly and completely within multiple coatings, at least one of which is impervious to moisture, to produce microcapsules which are insoluble in water at about
15 neutral pH, but which are soluble at acid pH. The microcapsules have a maximum particle size of about 1500 micrometers so they will stay suspended in an aqueous solution. The microcapsules can be tailored to have other features, such as successive layers of encapsulants having differing solubility
20 characteristics.

The microspheres are made using high speed rotation, e.g., a rotating disc method, that forms uniform, spherical particles of the required size. The microspheres are encapsulated with two or more coatings having differing
25 solubility characteristics. The resultant water impervious

microcapsul s can be admixed with aqueous solutions to form stable suspensions or mixtures that have a long shelf life in a concentration to provide dosage amounts of the drug that can be taken orally.

5

BRIEF DESCRIPTION OF THE DRAWING

Figure 1 is a cross sectional view of a high speed rotating disc system for preparing the microspheres of the invention.

10

Figure 2 is a cross sectional view of a centrifugal extrusion apparatus also used for preparing the microspheres and microcapsules of the invention.

Figure 3 is a cross sectional view of an air suspension chamber for encapsulating the microspheres of the invention.

DETAILED DESCRIPTION OF THE INVENTION

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The microspheres and microcapsules described in this invention can be prepared using a variety of methods, including but not limited to, a rotating disc system, spray drying, a centrifugal extrusion nozzle device, an air suspension cooler, phase separation and solvent evaporation. In the following paragraphs some of the procedures that can be used to prepare the microcapsules described hereinafter are set forth in detail. Variations will be known to those skilled in the art.

20

Microspheres of the desired particle size, i.e., less than 550 micrometers, and preferably between about 250 to 500

micrometers, can be made using a high speed rotating disc system as shown in Figure 1.

Referring to Figure 1, the high speed rotating disc system 10 comprises an emulsion feed tube 12 which is situated over and feeds onto a rotating disc 14. The disc 14 is rotated by means of a motor 16. The disc 14 and the motor 16 are enclosed in a chamber 18. The chamber 18 serves to dry or cool and solidify the microspheres and to collect them in a collection area 20. The chamber 18 is also fitted with a filter 22 and an exhaust system 24. The disc 14 is situated some distance above the collection area 20 to allow time for solidification of the microspheres.

In operation, a slurry of the solid drug particles, which are finely divided below about 100 micrometers, and a suitable matrix, e.g., a lipid or bioadhesive composition, is fed to the feed tube 12 and dropped onto the rotating disc 14. Droplets or microspheres are thrown out from the periphery of the disc 14 due to the centrifugal forces developed by the high speed rotation of the disc 14, and fall by gravity to the collection area 20 of the chamber 18. The small spherical liquid droplets or microspheres are dried/cooled and solidified during this free fall, and are then collected. If desired, the microspheres can be sieved to collect a desired particle size distribution. For a disc about 4 inches in diameter rotating at about 2000 rpm, and a drug particle size of about 50 micrometers, the

majority of the resultant microspheres have a particle size of about 105 to 500 micrometers. Control of the rotational speed of the disc 14 provides control over the size and size distribution of the microspheres.

5 Alternatively, and particularly suitably for hot melt matrix materials, microspheres can be formed in a centrifugal extrusion apparatus 110 as shown in Figure 2. Referring to Figure 2, the feed mixture or slurry of drug particles in the heated matrix material is fed through the center tube of a
10 concentric feed tube 112 by means of a seal (not shown) to a rotating head 114 fitted with one or more nozzles 116. A rotating shaft 118 rotates the head 114 at high speed by means of a motor 119. A mechanical stirrer (not shown) can be used in the feed tank (not shown) prior to pumping through the feed
15 tube 112 to provide continuous stirring of the drug particles in the hot melt matrix. As the head 114 rotates, the slurry of the drug in the matrix material flows through the inner orifice of the nozzle, creating a stream of the drug particles uniformly dispersed in the molten matrix material. This
20 extruded rod breaks into individual particles due to the high speed rotation of the head 114. The coated drug particles are cooled and solidify to form the desired microspheres as they free fall into the collection area 120. The size of the resultant microspheres can be controlled and is dependent upon
25 the feed rate, the speed of rotation of the head 114 and the

size of the nozzle openings. Microsphere particle sizes of from about 250 to 500 micrometers can be made readily.

In order to form the microcapsules of the invention, the microspheres as prepared above are encapsulated in one or more water impervious coatings.

The microcapsules can be prepared using the apparatus as shown in Figure 2, except using an additional feed line into the head 114. The slurry of the drug particles in the heated matrix material is fed through the center tube of the concentric feed tube 112. The desired encapsulant material and any additives are fed through a separate feed line to the outer tube 113 of the feed tube 112 to the head 114. The additional feed line is provided to supply the encapsulant material, e.g., a water impervious polymer dissolved in a solvent, as well as any optional materials desired such as colorants, flavorings, surfactants and other additives as desired. The slurry of the drug particles in the matrix material is fed along with the encapsulant material, to the head 114. The rotation of the head 114 causes the encapsulant or encapsulant mixture to flow through the outer orifice of the nozzle and the drug slurry to flow through the inner orifice of the nozzle, creating a rod of the drug slurry encased in a sheath of the encapsulant materials. If an organic solvent is used in the encapsulant, it is vaporized and can be collected through the exhaust (not shown) for disposal or recycled as desired.

The microcapsul s of the invention can also be mad in a modified air suspension coater apparatus as shown in Figure 3. The air suspension coater 210 comprises a chamber 212 fitted with an air distribution plate 216 through which an air atream
5 passes. A feed line 220 for supplying the encapsulant material is connected to a hydraulic or pneumatic nozzle 224 which atomizes the encapsulant material into small droplets which coat the microspheres.

In operation, the microspheres to be coated 226 are fed
10 through an entry port 228 in the chamber 212. An air stream supplied by means of a blower is fed into the chamber 212 and passes through the air distribution plate 216 to carry the microspheres smoothly past the head 224 for coating with encapsulant, and then up and over a coating partition 230 where
15 the coated microspheres are dried and cooled during free fall back outside the coating partition 230 to begin another cycle past the nozzle 224. The nozzle 224 is designed to atomize the coating material to allow a uniform, thin coating to be applied. In the course of multiple passes of the microspheres
20 past the nozzle 224, the encapsulant material uniformly coats the microspheres to the desired thickness. By changing the encapsulant feed to the nozzle 224, successive layers of desired encapsulant compositions are applied. Control of the air volume and temperature, atomizing conditions and rate of
25 application ensures a high degree of uniformity of the coatings

from batch to batch. Further, the closed system of the apparatus 210 provides excellent control of the conditions within the coater.

The drugs useful herein can be varied, but those particularly useful include antibacterial drugs including erythromycin or erythromycin ethyl succinate, and the penicillins, their salts, esters and hydrates, such as amoxicillin trihydrate, ampicillin trihydrate, penicillin V potassium and the like. These drugs are unstable in water. The drugs are supplied in solid form. If the drug is supplied in a larger particle size than desired, it can be milled to the appropriate particle size prior to coating with a lipid coating.

As used herein, the term lipid includes fatty acids, whether saturated or unsaturated, such as monobasic aliphatic carboxylic acids which form esters with glycerol or other alcohols to make fats, oils, waxes and other lipids. Also included are the esters and ethers of glycerides, the esters formed by reaction of the fatty acids with glycerol, such esters formed from pharmaceutically acceptable weak acids such as tartaric acid and its diacetyl derivative, acetic acid, ascorbic acid and citric acid, or one having a monophosphate group to yield the mono-phosphate ester. Suitable ethers are formed by reaction of the mono- or diglyceride with a functionally reactive lower alkyl, alkenyl, alkynyl, aryl or

substituted aryl group to produce the corresponding pharmaceutically acceptable ether, as is known in the art. Polyhydric alcohols such as octanol or a carbohydrate polyol, e.g., sucrose, are also useful in the present invention.

5 The matrix material can also be a bioadhesive polymer that will provide a delayed release of the drug. The bioadhesive polymer attaches to the stomach lining or mucin coating of the stomach, where it hydrates and is absorbed, thereby releasing the drug particles.

10 Suitable bioadhesive polymers include adhesive materials such as gelatin, polycarboxophil polymers, and Chitosan, commercially available from Protan of Norway. These matrix materials provide a delivery system which may provide a long acting dosage form by providing a reduced rate of emptying of the drug in the stomach, improve bioavailability of the drug, improve therapy, and increase the contact time of the drug in the desired absorption area.

 The microspheres are made into microcapsules by means of one or more enteric coatings. The enteric coatings can be tailored to have the drug absorbed in the body as desired, but at least one layer of enteric coating must be water insoluble at normal pH. Most antibiotics are meant to be absorbed in the intestinal tract, and thus must be protected from the high acid content gastric fluid of the stomach. Thus successive coatings may be insoluble in highly acid environments, i.e., pH below

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about 5, but soluble in less acid environments, i. . pH about 5.5 to 7.5 or higher.

Examples of known enteric coating materials useful herein include cellulose acetate phthalate, cellulose acetate trimellitate, hydroxypropyl methylcellulose phthalate, 5 polyvinyl acetate phthalate, shellac, methacrylic acid and methacrylic acid esters and zein. Partially hydrogenated vegetable oils, stearic acid, hydrogenated tallow triglycerides, food grade metal stearates, tallow and mixtures 10 thereof, and the like are used as lipids, carriers and modifiers for the microspheres. Suitable partially hydrogenated vegetable oil material is commercially available as Durkee 17 and Durkee KLX from Van den Burgh. Emersol 6349 stearic acid is commercially available from Emory Industries. Hydrogenated 15 tallow triglycerides are commercially available as Grocol 600-E from A. Gross and Co. Suitable tallow flakes are commercially available from Anderson Clayton/Hank Products, Inc. A series of methacrylic acid or methacrylic acid esters commercially available as Eudragit™ coatings, trademarks of 20 Rohm Pharma GmbH of Westerstadt, West Germany, have varying degrees of esterification, and are soluble at varying pH. Thus drugs which are meant to be absorbed in the small intestine will be encapsulated in a first enteric coating that is insoluble in water but soluble in acidic environment, e.g., the 25 stomach, and a second enteric coating which is insoluble in the

low pH gastric fluid of the stomach, but soluble in the less acid environment of the small intestine. For example, Eudragit™ E-100 is insoluble at neutral pH but soluble at a pH less than 5.5; Eudragit™ L-100-55 and L-30D are soluble at pH greater than 5.5. Metal stearates incorporated into these shell materials can increase water repellency. Other enteric coatings are known which are less soluble and can provide release of the drug over time, as will be known to those skilled in the art.

To provide both water impermeability, protection of the drug in the stomach and release in the intestine, and delayed release in the intestine, a series of enteric coatings will be applied to the microspheres. The outer layer will be water insoluble at normal pH, but will dissolve in the stomach (pH less than 5.5). The next layer will be insoluble at low pH in the stomach, but will dissolve and release the drug in the intestine (pH greater than 5.5). A third layer can provide delayed release until the drug enters the upper gastrointestinal tract, where the pH is higher again.

The enteric coatings are dissolved in an organic solvent, suitably one approved for medicinal use, such as acetone, methylene chloride, or the lower alcohols such as ethanol, or mixtures of such solvents, and applied to the microspheres as above. The organic solvents are removed by evaporation during the processing of the microcapsules.

The amount of coating applied to the microspheres is not critical, and can vary from 5 to 30% by weight of the microcapsule. It is important that the enteric coating be applied uniformly over the microsphere to ensure that the microspheres are protected from moisture, and that a given dosage of the drug will be released by the coatings at the appropriate portion and time in the digestive tract. Too thick a coating will delay dissolution of the coating and release of the drug.

The enteric coatings can also contain conventional additives such as suspending agents, emulsifying agents, essential oils, preservatives, flavoring or coloring agents and the like, as is known to one skilled in the art. Such additives can retain a desired texture, retard hydration or dehydration of the microsphere ingredients, and provide a uniform color and appearance.

The invention will be further described in the following examples, but the invention is not meant to be limited to the details thereof. In the examples, percent is by weight.

The actual erythromycin and erythromycin ethyl succinate content of the microspheres was determined by high pressure liquid chromatography (HPLC).

Actual amoxicillin trihydrate content in the microspheres was determined using an iodimetric titration method, see Code

of Federal Regulations: Food and Drugs, Vol 21, Chap 1 Part 436.204 (1988), 291.

EXAMPLE 1

Using the rotating disc apparatus of Figure 1, a slurry
5 containing 30.0% of Erythromycin USP, 44.1 % of Emersol 6349,
18.9% of Durkee 17 and 7.0% of aluminum stearate EA was heated
to 180°F and fed to the disc maintained at a temperature of
160°F.

The resultant microspheres had a particle size
10 distribution of 4.3% particles of less than 105 microns; 84.9%
particles of 105-250 microns; and 10.8% particles of 250-355
micrometers.

An assay of the microspheres having a particle size of
250-355 micrometers determined the actual Erythromycin content
15 to be 20.9%.

EXAMPLES 2-13

The procedure of Example 1 was followed except varying
the matrix composition and temperature. In these Examples,
30.0% of Erythromycin was employed. The data summarizing the
20 matrix composition, the matrix temperature, particle size and
weight % distribution obtained and the actual Erythromycin
content in the microspheres are summarized below in Table I. In
the Table, D17 represents Durkee 17; E6349 represents Emersol
6349; G 600-E represents Grocol 600-E; Zn St represents food
25 grade zinc stearate; Mg St represents food grade magnesium

stearate; Al St represents food grade aluminum st arate; and A
84K represents Atmul 84K.

TABLE I

<u>Example</u>	<u>Matrix Composition</u>	<u>Temp. °F</u>	<u>Particle Size, Micrometers</u>	<u>Weight %</u>	<u>Actual % Erythromycin</u>
2	70.0% D17	155	250-355	100	25.3
3	56.0% E6349 14.0% G600-E	155	105-250 250-355 355-500	18.4 68.4 13.2	25.7
4	52.5% E6349 17.5% D17	155	105-250 250-355	35.8 64.2	24.5
5	50.4% E6349 12.6% G600-E 7.0% Zn St	180	105-250 250-355	35.7 64.3	24.5
6	44.1% E6349 18.9% D17 7.0% Zn St	180	105-250 250-355	61.6 38.4	21.3
7	63.0% D17 7.0% Zn St	190	105-250 250-355	38.1 61.9	20.9
8	50.4% E6349 12.6% G600-E 7.0% Mg St	190	105-250 250-355	71.4 28.6	26.5
9	44.1% E6349 18.9% D17 7.0% Mg St	190	105-250 250-355	52.9 47.1	27.3

TABLE I
(Continued)

<u>Example</u>	<u>Matrix Composition</u>	<u>Temp. °F</u>	<u>Particle Size, Micrometers</u>	<u>Weight %</u>	<u>Actual % Erythromycin</u>
10	63.0% D17 7.0% Mg St	190-200	105-250	23.4	27.3
			250-355	55.8	
			355-500	20.8	
11	50.4% E6349 12.6% G600-E 7.0% Al St	180	105-250	83.0	21.3
			250-355	17.0	
12	63.0% D17 7.0% Al St	190-200	105-250	57.1	26.5
			250-355	23.1	
			355-500	19.8	
13	63.0% D17 7.0% A84K	190	150-250	25.8	27.3
			250-355	74.2	
Control	100% D17	155	<105	15.2	-
			105-250	58.7	
			250-355	26.1	

All of the microspheres were satisfactory.

EXAMPLE 14-22

- 5 The procedure for Examples 2-13 was followed except using 23% of Erythromycin. The data are summarized in Table II below, where the symbols are the same as for Table I.

TABLE II

<u>Example</u>	<u>Matrix Composition</u>	<u>Temp. °F</u>	<u>Particle Size, Micrometers</u>	<u>Weight %</u>
14	56.0% E6349	190	105-250	15.9
	14.0% G600-E		250-355	63.5
	7.0% Zn St		355-500	20.6
15	52.5% E6349	190	105-250	39.9
	17.5% D17		250-355	60.1
	7.0% Zn St			
16	70.0% D17	190	105-250	23.9
	7.0% Zn St		250-355	76.1
17	56.0% E6349	190	105-250	30.4
	14.0% G600-E		250-355	59.8
	7.0% Al St		355-500	9.8
18	52.5% E6349	190	105-250	67.5
	17.5% D17		250-355	32.5
	7.0% Al St			
19	70.0% D17	190-200	<105	8.2
	7.0% Al St		105-250	46.9
			250-355	44.9
20	56.0% E6349	190	105-250	67.0
	14.0% G600-E		250-355	33.0
	7.0% Mg St			
21	52.5% E6349	190	105-250	58.9
	17.5% D17		250-355	41.1
	7.0 Mg St			
22	70.0% D17	190	105-250	10.4
	7.0% Mg St		250-355	71.4
			355-500	18.2

All of the microspheres were satisfactory.

EXAMPLES 23-34

Using the apparatus of Figure 2 at a head speed of 2000 rpm, a head temperature of 190°F and a shell composition temperature of 180°F, a slurry of the drug in a matrix material was encapsulated with a single layer of various encapsulant compositions to produce microcapsules. Equal amounts by weight of the drug slurry and encapsulant were employed; thus the theoretical amount of Erythromycin was 15% in the microcapsules, which were sieved to collect a particle size of 250-500 micrometers for analysis. The data are summarized in Table III below where the symbols are the same as for Table I and N 060 represents Neutrene 060 triglycerides.

TABLE III

<u>Example</u>	<u>Encapsulant Composition</u>	<u>Slurry Composition</u>	<u>Actual % Erythromycin</u>
23	90% D17 10% Zn St	Ex3	8.7
24	80% E6349 20% G600-E	Ex3	9.5
25	80% E6349 20% G600-E	Ex2	5.4
26	100% D17	Ex2	6.1
27	90% D17 10% Zn St	Ex2	5.6
28	90% D17 10% Mg St	Ex2	6.1
29	100% D17	Ex3	-
30	90% D17 10% Mg St	Ex3	-
31	80% E6349 20% N060	56.0% E6349 14.0% N060 30% Erythromycin	6.7
32	100% D17	56.0% E6349 14.0% N060 30% Erythromycin	-
33	90% D17 10% Zn St	56.0% E6349 14.0% N060 30% Erythromycin	-
34	90% D17 10% Mg St	56.0% E6349 14.0% N060 30% Erythromycin	-

EXAMPLES 35-47

Using the apparatus of Figure 3, the microspheres prepared as in Examples 1-22 were encapsulated with two or more layers of various encapsulant compositions to produce microcapsules. The enteric coatings were dissolved in a solvent mixture, as shown, to which was optionally added a dyestuff, FD&C #1 Lake Blue, or FD&C #6 Lake Yellow. Eudragit™ E-100 provides stability in liquid suspension and in the mouth, (pH about 7) and is soluble in the stomach (pH less than 5.5). Eudragit™ L-100-55 will provide protection in the stomach and release the drug in the intestinal tract (pH over 5.5).

The data relating to the application of the coatings are summarized in Table IV below.

TABLE IV

Example	Microsphere Example	Shell Coating Composition % By Weight					Colorant #1	#6	Assay; Microcapsules
		Eudragit™ I-100-55	Eudragit™ E-100	Ethanol	Acetone	Methylene Chloride			
35	14	4	4	19.2 24.0	76.8	72.0	0.1	0.1	17.2
36	2	4	4	19.2 24.0	76.8 72.0		0.1	0.1	25.6
37	3	4	4	24.0 24.0	72.0	72.0	0.1	0.1	25.4
38	4	4	4	24.0 24.0	72.0	72.0	0.1	0.1	26.7
39	4	4	4	24.0 24.0	72.0 72.0		0.1	0.1	24.2
40	6	4	4	24.0 24.0	72.0 72.0		0.1	0.1	24.5
41	7	4	4	24.0 24.0	72.0 72.0		0.1	0.1	25.2
42	8	4	4	24.0 24.0	72.0 72.0		0.1	0.1	26.0
43	9	4	4	24.0 24.0	72.0 72.0		0.1	0.1	26.0
44	10	4	4	24.0 24.0	72.0 72.0		0.1	0.1	26.0
45	11	4	4	24.0 24.0	72.0 72.0		0.1	0.1	25.6
46	12	4	4	24.0 24.0	72.0 72.0		0.1	0.1	26.3
47	1	4	4	24.0 24.0	72.0 72.0		0.1	0.1	25.2

The data relating to the amount of the coatings applied is given below in Table V. where the result was obtained from a measurement of the amount of shell solution per weight of microsphere sample used during the coating process. The resultant microcapsules were in the 250 to 500 micrometer size range.

TABLE V

<u>Example</u>	<u>Weight % Shell</u>	
	<u>Inner</u>	<u>Outer</u>
36	14.0	13.9
43	13.0	13.3
44	13.0	13.8
45	13.0	13.2
46	13.0	13.1
47	13.0	13.6

EXAMPLE 48

Microcapsule samples made as in Examples 35-47 were evaluated for stability in simple syrup solution by storing for 1 and 4 weeks under accelerated shelf-life conditions at 37°C. The amount of antibiotic found in the syrup was determined. The data are summarized below in Table VI.

TABLE VI

<u>Microcapsules, Example</u>	<u>% Release 1 week</u>	<u>% Release 4 weeks</u>
35	<1.0*	6.2
36	0.11	2.4
37	<1.0*	2.5
38	<1.0*	2.5
39**	<1.0*	3.0
40	0.97	2.6
41	0.74	2.0
42	1.3	2.9
43	0.76	3.1
44	0.74	2.0
45	0.81	2.2
46	0.74	1.9
47**	0.64	1.6

* Estimated

** These samples were then shaken at 37°C for 48 hours. The % release for Example 39 was 3.6%; the % release for Example 47 was 1.8%.

It is appar nt that the microcapsules of the inv ntion are stable in aqueous solution for long periods of time, even under accelerated temperature conditions.

EXAMPLE 49

5 PART A. Microspheres were made following the procedure of Example 1 using 40.0% of erythromycin ethyl succinate within a 60% Durkee 17 matrix. The matrix temperature was 235°F.

The product contained 39.4% of particles 106-250 micrometers in size; and 60.4% of particles 250-355 micrometers
10 in size.

The actual amount of the antibiotic found in the microspheres was 31.2%.

PART B. The procedure of Example 35 was followed to make microcapsules of the microspheres of Part A. The first coating
15 was made using Eudragit™ L-30D and a second coating of Eudragit™ E-100 using a solvent mixture of 72.0 parts of methylene chloride and 24.0 parts of ethanol. The microcapsules contained 20.0% of erythrocmycin ethyl succinate.

EXAMPLES 50-55

20 The procedure of Example 1 was followed to prepare microspheres except that the drug employed was amoxicillin trihydrate in varying amounts. The data on microsphere preparation is summarized below in Table VII.

TABLE VII

<u>Example</u>	<u>Matrix Composition</u>	<u>Temp. °F</u>	<u>Particle Size, Micrometers</u>	<u>Weight %</u>	<u>% Amoxicillin Trihydrate Theoretical Found</u>
50	44.1% E 6349 18.9% D17 7.0% Al St	170	250-355	100	30.0 24.0
51	50.4% E 6349 12.6% Tallow flakes	200	106-250 250-355	35.9 64.1	37.0 Not-Tested
52	63.0 D17	200	250-355	100	37.0 32.5
53	63.0% D17	210	106-250 250-355	27.8 72.2	37.0 Not-Tested
54	68.0% D17	200	106-250 250-355	50.8 49.2	32.0 26.7
55	68.0% D17	200	250-355	100	32.0 21.8

EXAMPLES 56-60

The microspheres as in Examples 50-55 were encapsulated following the procedure of Example 35. The data relating to the encapsulations are summarized in Table VIII below:

TABLE VIII

Example	Microsphere Example	Shell Coatings, % By Weight				Solvent		Methylene Chloride	Colorant		Assay; Microcapsul
		Eudragit [™] L-100-55	Eudragit [™] E-100	Eudragit [™] L-30D		Ethanol	Acetone		#1	#6	
56	50	4	4			24.0 24.0	72.0 72.0		0.1	0.1	21.6
57	51	4				24.0	72.0		0.1		—
58	52		4	100		24.0		72.0	0.1		15.9
			4			24.0		72.0		0.1	
59	54		4			24.0		72.0			—
60	55		4	100		24.0		72.0	0.1	0.1	16.4
			4			24.0		72.0		0.1	3.1

EXAMPLE 61

The microcapsule samples made as in Example 56 using the microspheres as made in Example 50 were evaluated in simple syrup solution by storing for one, four and eleven weeks under accelerated shelf life conditions at 37°C. The amounts of antibiotic found in the syrup after vigorous shaking is as follows: 0.0% release after one week; 3.1% release after four weeks and 26.5% release after eleven weeks.

It is apparent that the microspheres of the invention can be encapsulated with multiple coatings as desired using the processes of the invention. The resultant microcapsules are small in size, are water impervious, and can be tailored so that the drug is protected in aqueous solutions, and can be released as desired, and when desired, to optimize the drug's effectiveness. The invention also provides a means of dispensing a water unstable drug in an aqueous solution that is stable and which can be stored at room temperature for extended periods of time.

We claim:

- 5 1. A microcapsule comprising a microsphere core particle comprising a solid drug particle in a matrix of a lipid or bioadhesive polymer encapsulated in two or more enteric coatings, at least one coating being impervious to water, said microcapsules having a particle size of 1500 micrometers or less.
- 10 2. A microcapsule according to claim 1 wherein said drug is selected from the group consisting of erythromycin or erythromycin ethyl succinate.
- 15 3. A microcapsule according to claim 1 wherein said drug is a penicillin.
- 20 4. A microcapsule according to claim 3 wherein said drug is selected from the group consisting of ampicillin, amoxicillin, and penicillin V, their salts and hydrates.
5. A microcapsule according to claim 1 wherein said matrix is a lipid.
- 25 6. A microcapsule according to claim 1 wherein said matrix is a bioadhesive polymer.

7. A microcapsule according to claim 1 wherein said one coating is insoluble in water at neutral pH and soluble in fluid having a pH of less than 5.5.
- 5 8. A microcapsule according to claim 7 wherein a second coating is soluble in fluid having a pH of more than 5.5.
9. A microcapsule according to claim 1 wherein the microsphere core particle comprises a solid drug particle in a bioadhesive
- 10 polymer matrix encapsulated in two or more enteric coatings, at least the outer coating being impervious to water.
10. A microcapsule according to claim 1 having a particle size of from about 250 to about 550 micrometers.
- 15 11. A microcapsule according to claim 1 wherein said microsphere has a particle size of from about 105 to about 500 micrometers.
- 20 12. A microcapsule according to claim 1 wherein said microsphere has a particle size of from about 250 to about 355 micrometers.

13. A method of making microspheres which comprises .

5 a) admixing solid drug particles having a particle size of less than 500 micrometers with a matrix of a lipid or a bioadhesive polymer to form a slurry,

b) forming microspheres having a drug core and a matrix coating thereon having a particle size of between about 105 to 500 micrometers, and

c) solidifying said microspheres.

10

14. A method according to claim 13 wherein said forming comprises dropping said slurry onto a disc rotating at high speed whereby said slurry is divided into droplets.

15

15. A method according to claim 13 wherein said forming comprises extruding a drug particle slurry in a matrix material of a lipid or bioadhesive polymer through a nozzle rotating at high speed.

20

16. A method of making microcapsules which comprises encapsulating the microspheres of claim 10 with two or more enteric coatings, at least the outer one of which is water insoluble.

25

17. A method according to claim 16 wherein the outer coating is water insoluble at neutral pH but is soluble in a fluid at a pH

below 5.5 and the inner coating is soluble in a fluid having a pH over 5.5.

18. A stable medication comprising a sugar syrup containing a
5 suspension of the microcapsules of claim 1.

FIG. 1

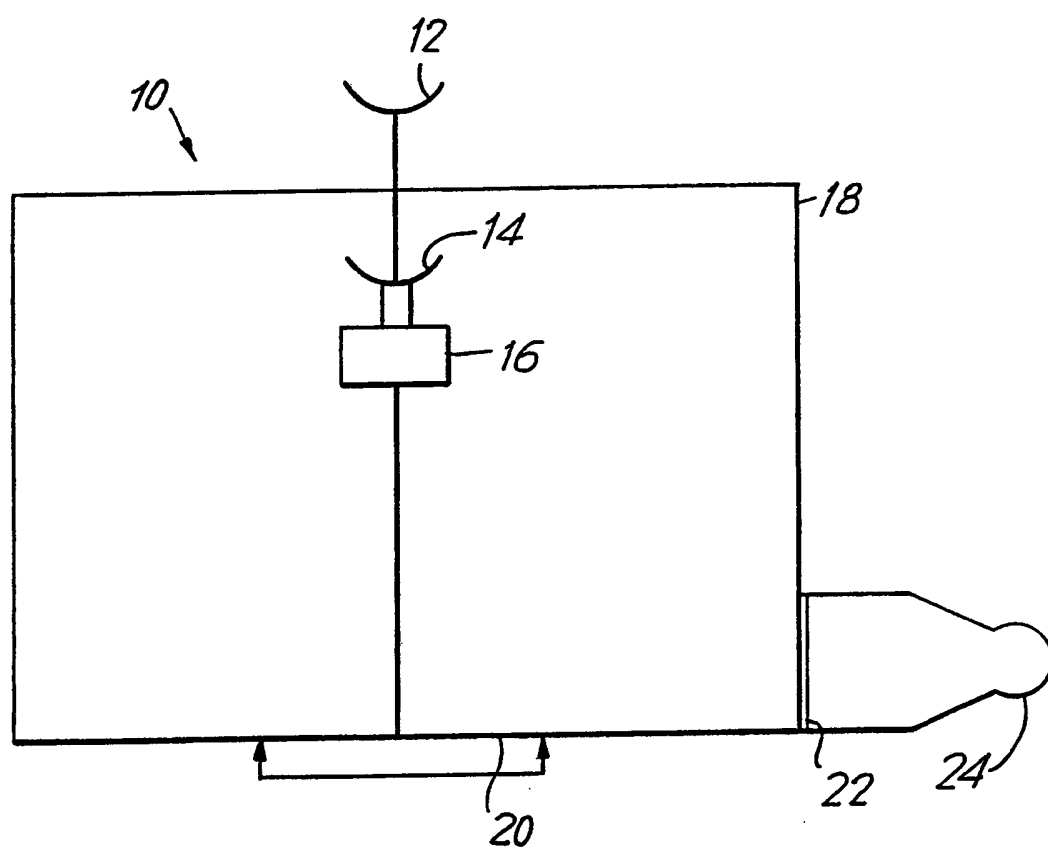
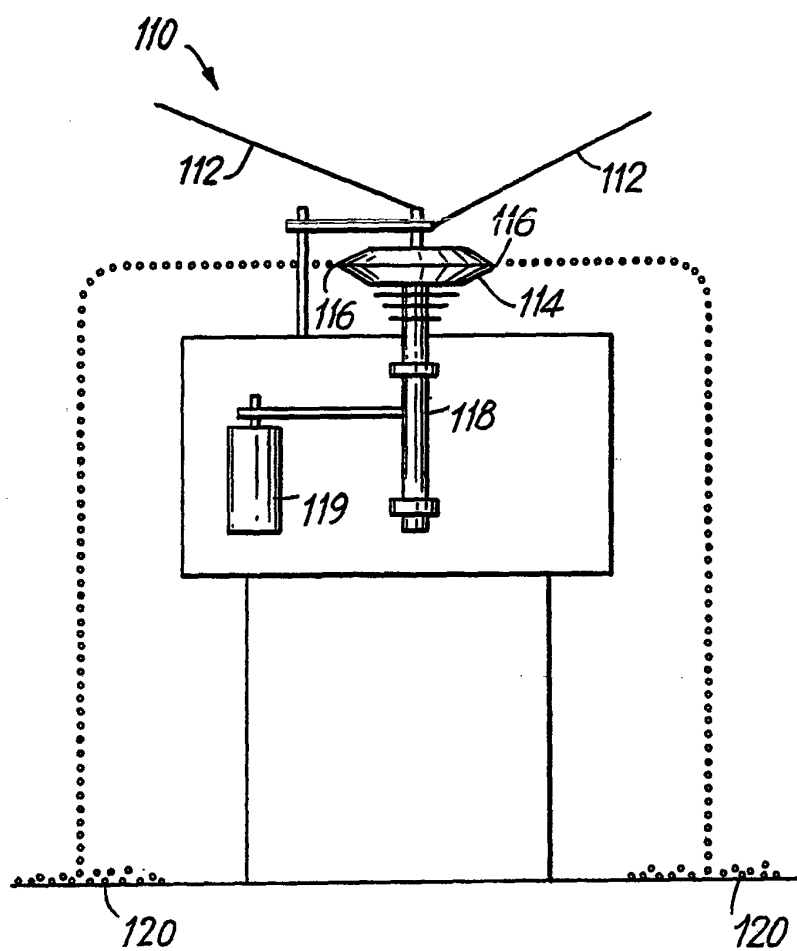


FIG. 2



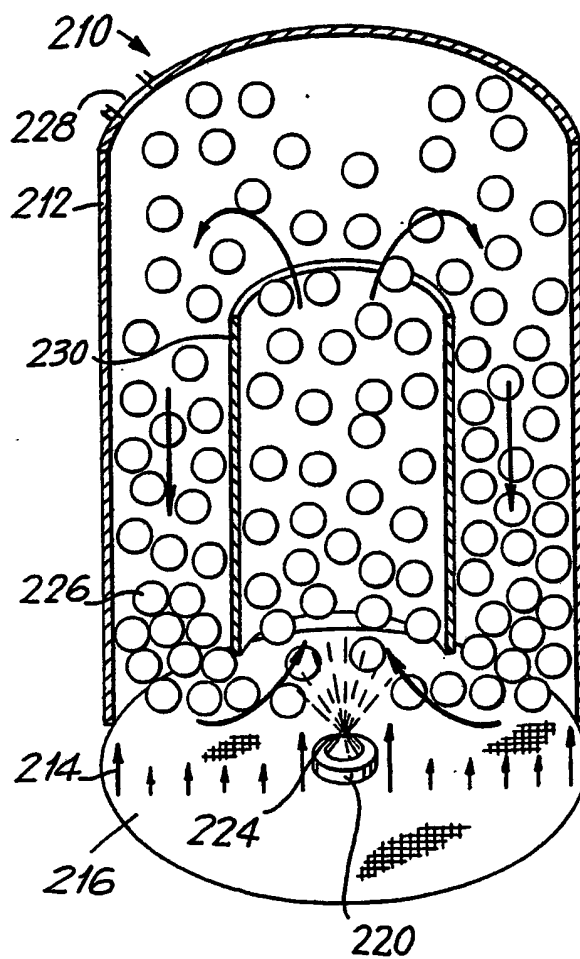


FIG. 3

INTERNATIONAL SEARCH REPORT

International Application No. PCT/US91/04198

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) *		
According to International Patent Classification (IPC) or to both National Classification and IPC IPC(5): A61K 9/16 U.S. CL. 424/497		
II. FIELDS SEARCHED		
Minimum Documentation Searched ⁷		
Classification System	Classification Symbols	
U.S.	424/497, 440, 494,	
Documentation Searched other than Minimum Documentation to the extent that such Documents are included in the Fields Searched *		
III. DOCUMENTS CONSIDERED TO BE RELEVANT *		
Category *	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³
X	US, A, 4,849,227 (CHO) 18 JULY 1989; See See column 5, lines 42-46.	1
Y	US, A, 4,250,166 (MAEKAWA) 10 FEBRUARY 1981; See entire document.	7, 8, 13-18
Y	US, A, 4,713,247 (SAKAMOTO) 15 DECEMBER 1987; See entire document.	7, 8, 13-18
<p>* Special categories of cited documents: ¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"A" document member of the same patent family.</p>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search	Date of Mailing of this International Search Report	
15 August 1991	10 OCT 1991	
International Searching Authority	Signature of Authorized Officer	
ISA/US	D. Gabrielle Phelan	